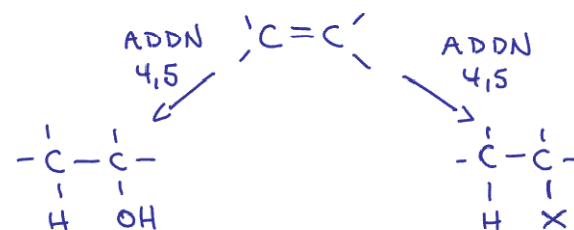


Survey of Key Functional Groups

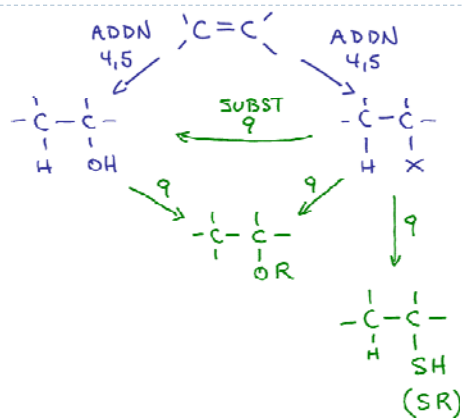
Ch 8.1

Overview



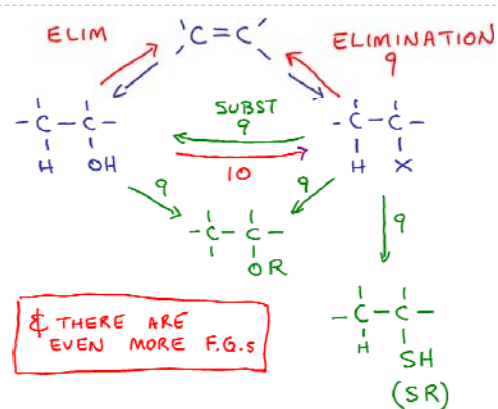
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Overview



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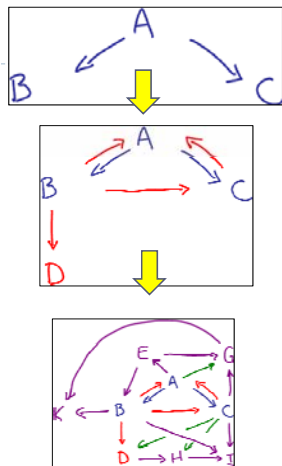
Overview



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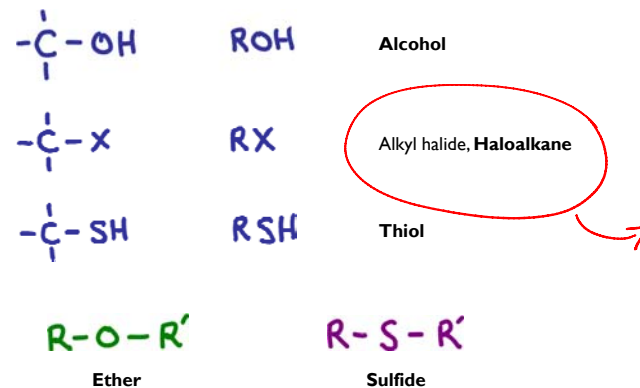
Concerns

- ▶ Getting overwhelmed
 - ▶ Strange new formulas
 - ▶ Strange new names
 - ▶ Strange new reagents
 - ▶ Many new relationships
 - ▶ leads to **brain meltdown !!!**
- ▶ Info builds ONE step at a time
- ▶ Ch 8 surveys FG
 - ▶ Names, drawings, properties
 - ▶ No new reactions (almost)



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Functional Group Names



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RX Names

Common or Substitutive (IUPAC)?

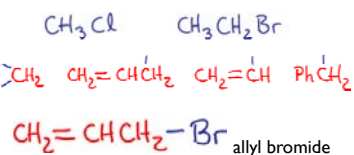
- ▶ **Alkyl halide**
 - ▶ Alkyl group attached
 - ▶ to **Halogen**

- ▶ Methyl chloride
- ▶ Ethyl bromide

- ▶ Special compounds
- ▶ Special groups
 - ▶ methylene, allyl, vinyl, benzyl
- ▶ Special history

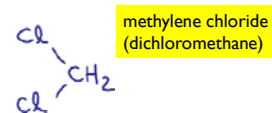
- ▶ **Haloalkane**
 - ▶ **Halogen** substituent
 - ▶ attached to **Alkane**

- ▶ Chloromethane
- ▶ Bromoethane



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Special Compounds Use Common Names

methylene chloride
(dichloromethane)

Haloforms

chloroform CHCl_3
 bromoform CHBr_3
 iodoform CHI_3

carbon tetrachloride



carbon tetrabromide



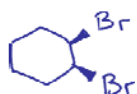
Useful solvents

- hydrophobic
- moderately polar
- volatile
- but all linked to health problems**

these slides will be posted

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Give "haloalkane" (IUPAC) names

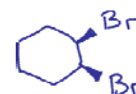


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Give "haloalkane" (IUPAC) names



2-fluoropropane



cis-1,2-dibromocyclohexane

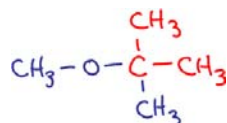
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ROR' Names
Common & Substitutive

▶ **Dialkyl ether**

- ▶ 2 **Alkyl** groups attached
- ▶ to oxygen make **Ether**

- ▶ Diethyl ether ("ether")
- ▶ Methyl *tert*-butyl ether (MTBE)

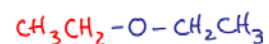


▶ **Alkoxyalkane**

- ▶ **Alkoxy** (RO) substituent
- ▶ attached to **Alkane** (R')

- ▶ **Alkoxy** = alkyl (R) + oxy
- ▶ **Alkane** = alkyl (R') + ane

- ▶ Ethoxyethane
- ▶ 2-Methoxy-2-methylpropane

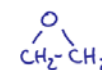


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ROR' Names
Additional considerations



Many special rings



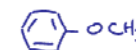
oxirane
epoxide
"alkene" oxide

Tetrahydrofuran
(THF)



1,4-dioxane

furan



anisole

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RSR' Names Common & Substitutive

▶ Dialkyl sulfide

- ▶ 2 Alkyl groups attached
- ▶ to sulfur make **Sulfide**

ethyl methyl sulfide



▶ (Alkylthio)alkane

- ▶ Alkylthio (RS) substituent
- ▶ attached to Alkane (R')

▶ Alkylthio = alkyl (R) + thio

▶ Alkane = alkyl (R') + ane

(methylthio)ethane

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RX, ROR', RSR' name similarities

▶ Common names used widely

- ▶ Special compounds (names often break “common” rules)
- ▶ Rings (ROR, RSR only)

▶ IUPAC

- ▶ Substituted alkane
 - ▶ Haloalkane
 - ▶ Alkoxyalkane
 - ▶ (Alkylthio)alkane

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ROH, RSH name similarities

▶ Common names used

- ▶ Special compounds

▶ IUPAC

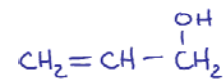
- ▶ New suffix replaces “alkane”
- ▶ ROH, **alkanol**
- ▶ RSH, **alkanethiol**

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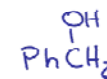
Alcohol names - common or IUPAC?

▶ Common names are fading out

- ▶ Special groups

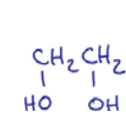


allyl alcohol

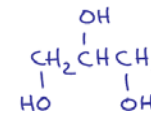


benzyl alcohol

- ▶ Special compounds



ethylene glycol



glycerol

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IUPAC - Alkanol

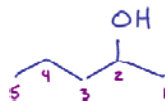
- ▶ Parent chain = longest chain bearing OH

- ▶ Alkanol = alkane + ol

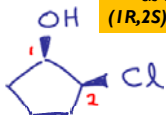
} Parent = alkanol

- ▶ # = position of OH

- ▶ give smallest # possible
 - ▶ always #1 in ring

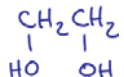
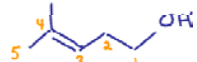


2-pentanol
pentan-2-ol



2-chlorocyclopentanol
cis-2-chlorocyclopentanol
(1*R*,2*S*)-2-chlorocyclopentanol

4-methyl-3-penten-1-ol



1,2-ethanediol
ethane-1,2-diol

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IUPAC - Alkanethiol

- ▶ Parent chain = longest chain bearing SH

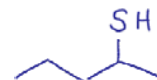
- ▶ Keep 'e' from alkane

- ▶ Add 'thiol' (one word)

} Parent = alkanethiol

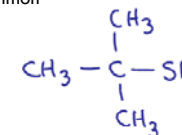
- ▶ # = position of SH

- ▶ give smallest # possible
 - ▶ always #1 in ring



2-pentanethiol
pentane-2-thiol

Common



tert-butyl mercaptan

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When groups compete: alcohol vs. thiol

- ▶ How would you name this?



- ▶ 5 C chain. Could be...

- ▶ some kind of 1-pentanol
 - ▶ some kind of 2-pentanethiol

- ▶ Who wins? OH or SH?

- ▶ And how do you name other group as substituent?

- ▶ IUPAC Rules !!!

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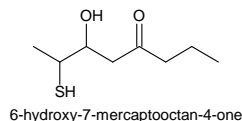
IUPAC Rules (p. 327)

1. Identify the **principal group**
 - a. Rank candidates (OH beats SH)
2. Identify the **principal chain**
 - a. Chain contains principal group(s)
 - b. Chain contains max. number double & triple bonds, etc.
3. Number the **principal chain**
 - a. Lowest # for C bearing principal group
 - b. Lowest # for multiple bonds, etc.
4. Construct name
 - a. Name principal chain, group, and #
 - b. Name substituents and #'s

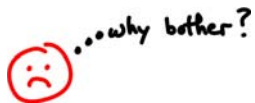
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But IUPAC rules are problematic

- ▶ Complicated
 - ▶ Must fill in **etc.** (several rules)
 - ▶ Must know principal group **rankings**
 - ▶ Must know **substituent names**



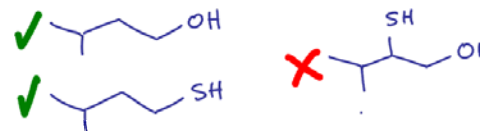
- ▶ And IUPAC names are not even universally accepted



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Chem 201 simplification

- ▶ I will only give you **naming problems** for compounds that contain **ONE** principal group
 - ▶ You don't need to learn principal group **rankings**
 - ▶ You don't need to learn **substituent names**



- ▶ But ... I will expect you to be able to work with more complicated names **when I provide them**

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Typical naming problems

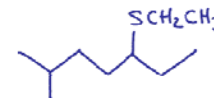
- ▶ Draw 5-(ethylthio)-2-methylheptane

- ▶ Draw an oxirane

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Typical naming problems

- ▶ Draw 5-(ethylthio)-2-methylheptane



- ▶ Draw an oxirane



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